

INFORMATION TO USERS

The most advanced technology has been used to photograph and reproduce this manuscript from the microfilm master. UMI films the text directly from the original or copy submitted. Thus, some thesis and dissertation copies are in typewriter face, while others may be from any type of computer printer.

The quality of this reproduction is dependent upon the quality of the copy submitted. Broken or indistinct print, colored or poor quality illustrations and photographs, print bleedthrough, substandard margins, and improper alignment can adversely affect reproduction.

In the unlikely event that the author did not send UMI a complete manuscript and there are missing pages, these will be noted. Also, if unauthorized copyright material had to be removed, a note will indicate the deletion.

Oversize materials (e.g., maps, drawings, charts) are reproduced by sectioning the original, beginning at the upper left-hand corner and continuing from left to right in equal sections with small overlaps. Each original is also photographed in one exposure and is included in reduced form at the back of the book. These are also available as one exposure on a standard 35mm slide or as a 17" x 23" black and white photographic print for an additional charge.

Photographs included in the original manuscript have been reproduced xerographically in this copy. Higher quality 6" x 9" black and white photographic prints are available for any photographs or illustrations appearing in this copy for an additional charge. Contact UMI directly to order.

U·M·I

University Microfilms International
A Bell & Howell Information Company
300 North Zeeb Road, Ann Arbor, MI 48106-1346 USA
313/761-4700 800/521-0600

PREVIEW

Order Number 8904479

**Enzyme studies and salt sensitive mutant characterization in
relation to the halotolerance of *Halomonas elongata***

Bylund, James Eugene, Ph.D.

The University of Nebraska - Lincoln, 1988

U·M·I

300 N. Zeeb Rd.
Ann Arbor, MI 48106

PREVIEW

ENZYME STUDIES AND SALT SENSITIVE MUTANT CHARACTERIZATION
IN RELATION TO THE HALOTOLERANCE OF
HALOMONAS ELONGATA

by

James E. Bylund

A DISSERTATION

Presented to the Faculty of
The Graduate College in the University of Nebraska
In Partial Fulfillment of Requirements
For the Degree of Doctor of Philosophy
Major: Biological Sciences

Under the Supervision of Professor Eugene L. Martin

Lincoln, Nebraska

December, 1988

TITLE

Enzyme Studies and Salt Sensitive Mutant Characterization in

Relation to the Halotolerance of Halomonas elongata.

BY

James E. Bylund

APPROVED

DATE

Dr. Eugene L. Martin

9-19-88

Dr. Thomas L. Thompson

9-19-88

Dr. John K. Dyer

9-19-88

Dr. John Osterman

9-19-88

Dr. Lloyd B. Bullerman

9-19-88

SUPERVISORY COMMITTEE

GRADUATE COLLEGE

UNIVERSITY OF NEBRASKA

ENZYME STUDIES AND SALT SENSITIVE MUTANT CHARACTERIZATION
IN RELATION TO THE HALOTOLERANCE OF
HALOMONAS ELONGATA

James E. Bylund, Ph.D.

University of Nebraska-Lincoln, 1988

Advisor: Eugene L. Martin

The cytoplasmic enzyme alanine dehydrogenase (EC 1.4.1.1) and the extracellular enzymes alkaline and acid phosphatases (EC 3.1.3.1 and EC 3.1.3.2, respectively) from Halomonas elongata were partially purified from cells grown in an L-alanine defined medium with NaCl at 0.05, 1.37 and 3.4 M.

The activity profile for alanine dehydrogenase isolated from the three cell samples showed optimal activity at a NaCl concentration corresponding to the internal sodium concentration for each sample. Enzyme isolated from 0.05 M NaCl cells had highest activity at 0-20 mM NaCl; while enzyme isolated from 1.37 and 3.4 M NaCl cells had optimal activity at 340 and 500-600 mM NaCl respectively. Enzyme activity decreased when KCl or LiCl was substituted for NaCl. Polyacrylamide gel electrophoresis followed by histochemical staining revealed that 0.05 and 1.37 M NaCl cell extracts had two bands of activity (isozymes), while the 3.4 M NaCl cell extracts exhibited a single activity band.

Alkaline (AlP) and acid (AcP) phosphatases were cytochemically localized on the cell envelope. Enzyme assays were conducted at pH 9.0 (AlP) and 5.0 (AcP) with varying

concentrations of NaCl, KCl or LiCl in the assay buffer. Results show higher acid phosphatase activity compared to that of alkaline phosphatase and all enzyme activities were optimal at NaCl concentrations similar to the medium NaCl concentrations in which the cells were grown. However, all samples demonstrated strong activities at 4.0 M NaCl and at some KCl concentrations. Enzyme activities decreased significantly (with exceptions) when KCl or LiCl was substituted for NaCl. Polyacrylamide gel electrophoresis followed by histochemical staining for the phosphatases showed only one band for both enzymes from each cell sample grown at different NaCl concentrations.

Salt sensitive mutants of *H. elongata* were generated by frame shift mutagens and isolated by selecting for the inability to grow at high NaCl concentrations. Salt tolerance studies revealed that each mutant tested had the same salt tolerance profile. The salt tolerance of one mutant (Hm1) could be extended by the presence of large amounts of tryptophan and glycerol. Tryptophan pathway studies demonstrated that Hm1 had a salt sensitive conditional mutation in the trp operon, possibly in the trp B gene region. Transport experiments using [¹⁴C]tryptophan and [¹⁴C]aminoisobutyric acid showed that Hm1 had a diminished uptake of these compounds compared to the parent.

ACKNOWLEDGEMENTS

The completion of this dissertation has come about through the efforts of several people whom I would like to acknowledge at this time. First and foremost I would like to thank my advisor Eugene Martin for his guidance, help and friendship. His advice and discussion have been invaluable to me during my time here.

Special thanks are also due to Thomas Thompson for initially sparking my interest in microbiology and for his continued support and guidance, Jack Dyer for his polyacrylamide gel electrophoresis work and unselfish commitment to this project, Lloyd Bullerman and John Osterman whose advice and discussion is most appreciated, Nancy Van Pelt for her friendship and patience during the writing of this work, Dwayne Wiley, without whose encouragement I wouldn't be here, Ellen Jensen, John David, Mike Black, Mike Heaton, and Ben Brenton for their willing assistance and helpful discussions.

Finally I would like to thank my family and especially my mother for her continued support and belief in me even when I didn't know what I was doing.

TABLE OF CONTENTS

	Page
ACKNOWLEDGMENTS	ii
INTRODUCTION	1
LITERATURE REVIEW	3
MATERIALS AND METHODS	25
Growth Media and Culture Conditions	25
Purification of Enzymes	25
Alanine Dehydrogenase Assay	26
Specific Activity and Protein Concentration Determination for Alanine Dehydrogenase	27
Electrophoresis and Histochemical Staining of Alanine Dehydrogenase	27
Heat Inactivation of Isozymes	28
Alkaline and Acid Phosphatase Assay	28
Specific Activity and Protein Concentration Determination of Alkaline and Acid Phosphatase	29
Electrophoresis and Histochemical Staining of Alkaline and Acid Phosphatase	30
Enzyme Cytochemistry	30
Kinase Effector Growth Study	31
Mutant Isolation	32
Mutant Salt Tolerance	32
Tryptophan Pathway Studies	33
Effects of Glycerol	35
Transport Studies	35
RESULTS	38
Alanine Dehydrogenase Salt Tolerance	38

	Page
Effects of Monovalent Cations on Alanine Dehydrogenase	41
Electrophoresis of Alanine Dehydrogenase	46
Heat Inactivation	46
Alkaline and Acid Phosphatase Activity	46
Alkaline and Acid Phosphatase Salt Tolerance	52
Effect of Monovalent Cations on Alkaline and Acid Phosphatase	54
Electrophoresis of Alkaline and Acid Phosphatase	60
Cytochemical Localization	67
Kinase Effector Study	67
Mutant Isolation	67
Mutant Salt Tolerance	71
Tryptophan Pathway Studies	74
Effects of Glycerol	89
Transport Profiles	95
DISCUSSION	98
LITERATURE CITED	115

LIST OF TABLES

Table		Page
1	Salt response of different microorganisms	4
2	Comparison of intracellular Na ⁺ and K ⁺ concentrations in various halophilic and halotolerant bacteria	10
3	Purification of alanine dehydrogenase	39
4	Effects of Na ⁺ , K ⁺ and Li ⁺ on specific activity of alanine dehydrogenase	40
5	Effects of Na ⁺ , K ⁺ and Li ⁺ on specific activity of alkaline phosphatase	50
6	Effects of Na ⁺ , K ⁺ and Li ⁺ on specific activity of acid phosphatase	51
7	Effects of amino acids and glycerol on the salt tolerance of Hm1	76
8	Effects of amino acids and glucose on the salt tolerance of Hm1	77
9	Effects of amino acids and sucrose on the salt tolerance of Hm1	78
10	Effects of amino acids and mannitol/ mannose on the salt tolerance of Hm1	79
11	Effects of amino acids and succinate/ -ketoglutarate on the salt tolerance of Hm1	80
12	Tryptophan pathway studies on Hm1 at various NaCl concentrations in defined medium	90

LIST OF FIGURES

Figure	Page
1 Examples of organic osmotica	8
2 Effect of NaCl on enzymes from extreme, moderate and non-halophiles	20
3 Effect of NaCl on alanine dehydrogenase from cells grown at 0.05, 1.37 and 3.4 M NaCl	42
4 Effect of NaCl, KCl and LiCl on alanine dehydrogenase from cells grown at 0.05 M NaCl	43
5 Effect of NaCl, KCl and LiCl on alanine dehydrogenase from cells grown at 1.37 M NaCl	44
6 Effect of NaCl, KCl and LiCl on alanine dehydrogenase from cells grown at 3.4 M NaCl	45
7 Histochemical staining for alanine dehydrogenase on a polyacrylamide gel electrophoretogram of extracts from cells grown at 3.4, 1.37 and 0.05 M NaCl	47
8 Histochemical staining for alanine dehydrogenase on a polyacrylamide gel electrophoretogram from cells grown in alanine defined medium and glutamate defined medium at 1.37 M NaCl	48
9 Histochemical staining for alanine dehydrogenase on a polyacrylamide gel electrophoretogram from column extracts preincubated from 40-80°C from cells grown at 1.37 and 0.05 M NaCl	49
10 Effect of NaCl on alkaline phosphatase from cells grown at 0.05, 1.37 and 3.4 M NaCl	53
11 Effect of NaCl on acid phosphatase from cells grown at 0.05, 1.37 and 3.4 M NaCl	55

Figure		Page
12	Effect of NaCl, KCl and LiCl on alkaline phosphatase from cells grown at 0.05 M NaCl	56
13	Effect of NaCl, KCl and LiCl on acid phosphatase from cells grown at 1.37 M NaCl	57
14	Effect of NaCl, KCl and LiCl on alkaline phosphatase from cells grown at 1.37 M NaCl	58
15	Effect of NaCl, KCl and LiCl on acid phosphatase from cells grown at 1.37 M NaCl	59
16	Effect of NaCl, KCl and LiCl on alkaline phosphatase from cells grown at 3.4 M NaCl	61
17	Effect of NaCl, KCl and LiCl on acid phosphatase from cells grown at 3.4 M NaCl	62
18	Histochemical staining for alkaline phosphatase on a polyacrylamide gel electrophoretogram of extracts from cells grown at 0.05 and 3.4 M NaCl	63
19	Histochemical staining for alkaline phosphatase on a ployacrylamide gel electrophoretogram of extracts from cells grown in alanine and glutamate defined mediums at 1.37 M NaCl	64
20	Histochemical staining for acid phosphatase on a polyacrylamide gel electrophoretogram of extracts from cells grown at 0.05 and 3.4 M NaCl	65
21	Histochemical staining for acid phosphatase on a polyacrylamide gel electrophoretogram of extracts from cells grown in alanine and glutamate defined mediums at 1.37 M NaCl	66
22	Enzyme cytochemistry	68
23	Effect of kinase stimulator and inhibitor on the growth of 1H9 at 0.1 and 2.0 M NaCl	69
24	Effect of kinase stimulators and an inhibitor on the growth of Hm1 at 0.1 and 2.0 M NaCl	70

Figure		Page
25	Microscope slide photographs of 1H9 and Hm1 log and stationary phase cells	72
26	Effect of NaCl on the growth of salt sensitive mutants	73
27	Effect of tryptophan and glycerol on the growth of Hm1 at 1.37 M NaCl	75
28	Synthesis of tryptophan from anthranilate	81
29	Effect of tryptophan and trptophan intermediates on the growth of Hm1 at 1.37 M NaCl	82
30	Chemical structure of glyphosate	84
31	Effect of glyphosate on the growth of Hm1 at 0.0 and 0.375 M NaCl	85
32	Effect of glyphosate on the growth of 1H9 at 0.375 M NaCl	86
33	Effect of glyphisate on the growth of 1H9 at 1.37 M NaCl	87
34	Effect of glyphosate on the growth of 1H9 at 3.4 M NaCl	88
35	Scanning UV spectrograph of 1H9 extracts from cells grown at 1.37 and 3.4 M NaCl	89
36	Effect of tryptophan concentration on the growth of Hm1 in a defined medium at 0.7 M NaCl	92
37	Effect of glycerol on the growth of 1H9 at 4.5 and 5.0 M NaCl	93
38	Effect of glycerol, glucose and sucrose on the growth of 1H9 at 4.5 M NaCl	94
39	Effect of NaCl on the uptake of AIB by 1H9 and Hm1	96
40	Effect of NaCl on the uptake of tryptophan by 1H9 and Hm1	97
41	Chemical structure of ectoin and tryptophan	109

Figure	Page
42 Aromatic amino acid biosynthesis pathway	110

PREVIEW

INTRODUCTION

A physiological explanation of halotolerance of bacteria has been the subject of research for many years. Available studies lead one to believe that the mechanism of halotolerance is not due to one factor but rather to the sum total of the properties of various cell components. If these factors are numerous and complex in nature, then a true understanding of halotolerance may be difficult to achieve. However, if the underlying mechanisms of halotolerance are relatively few, then they may be easier to understand and to control.

Bacteria of the genus Halomonas were originally isolated from a solar salt facility on the island of Bonaire in the Netherlands Antilles. H. elongata was found throughout the facility, demonstrating an extensive tolerance to salt. Since this organism has the ability to grow over an extended salt range and has a nutritional requirement for sodium, Vreeland and Martin (1980) designated H. elongata a halotolerant bacterium. Investigations into the halotolerance of Halomonas include the internal solute composition (Vreeland et al., 1983), aminoisobutyric acid transport (Martin et al., 1983), and cell wall and phospholipid composition (Vreeland et al., 1984).

The deleterious effects of monovalent salts on biological macromolecules are well known. A study of the

2

effects of salts on enzymes of Halomonas would provide further insight into the mechanism of halotolerance of this organism. A conventional and potentially profitable approach to the question of salt tolerance would be the study of mutants having diminished salt tolerance. Characterization of such mutants could answer key questions concerning the halotolerant mechanisms of Halomonas.

This dissertation: (1) examines and compares the salt response profiles of the intracellular enzyme alanine dehydrogenase and the extracellular enzymes alkaline and acid phosphatase of H. elongata, and (2) explores the physiological characteristics of a salt sensitive mutant of H. elongata.

PREVIEW

LITERATURE REVIEW

In microbiology sodium chloride at significant concentrations is usually thought of as an unhealthy, stressful environment for microorganisms. However microbiologists have found such salty environments as ancient salterns, the Great Salt Lake, the (not quite) Dead Sea, and other hypersaline systems the domain of halophilic and halotolerant microorganisms. These organisms, adapted to tolerate NaCl or to require it as a nutrient (Larsen, 1962), represent a diverse group of yeasts, molds, algae, cyanophytes, and bacteria. The tolerance range or requirement that an organism has for salt has been the basis for many classification schemes.

The first workable scheme (Larsen, 1962) categorized microorganisms into four groups based upon their increasing requirements for NaCl: nonhalophiles, slight (marine), moderate, and extreme halophiles. Obligate halophiles require salt to survive and facultative halophiles do not require salt, but are stimulated by its presence.

Kushner (1978, 1985) attempted to associate molar values with the adjectives slight, moderate, and extreme (Table 1). The problems arising from using this classification scheme is that many organisms overlap the boundaries between classifications depending on the medium composition and the temperature at which they are grown (Kushner, 1978). Researchers have also induced algae that

TABLE 1.

a

4

Salt Response of Different Microorganisms

Category	Reaction	Examples
Nonhalophile	Grows best in less than 0.2 M salt	Most eubacteria and freshwater organisms
Slight halophile	Grows best in 0.2-0.5 M salt	Marine organisms
Moderate halophile	Grows best in 0.5-2.5 M salt. Organisms able to grow in < 0.1 M salt are facultative halophiles	<u>Vibrio costicola</u> <u>Paracoccus halodenitrificans</u>
Borderline extreme halophile	Grows best in 1.5-4.0 M salt	<u>Ectothiorhodospira halophila</u> <u>Halobacterium volcanii</u>
Extreme halophile	Grows best in 2.5-5.2 M salt	<u>Halobacterium salinarium</u> <u>Halococcus morrhuae</u>
Halotolerant	Nonhalophile that can tolerate salt If growth range extends above 2.5 M salt, it is extremely halotolerant	<u>Halomonas elongata</u> Solute-tolerant yeasts, molds and algae

a

Adapted from Table 3 in Kushner (1978).

would not grow at high salt concentrations to thrive there; and others, that once needed higher salt, to grow at more moderate levels (Borowitzka et al., 1977), thus demonstrating that some organisms can change their response to salt.

Vreeland (1987) presents a different view of halotolerance and halophilism in microorganisms. This classification scheme is independent of the environmental salt concentration at which the organism resides. This classification scheme is, instead, based on the differences between upper and lower limits of growth. Though this would seem to be an answer to the problems stated earlier there is still the question of halophilism vs. halotolerance.

Reed (1986) defines halophiles as organisms that have an unusual requirement for salt which separates them from other organisms; any organism that has a minimum requirement for salt greater than 0.5 M can be regarded as a halophile. Halotolerant organisms are defined as having the capability of sustained growth at a level of 1.0 M NaCl or greater, but no absolute requirement for salt in amounts greater than 0.5 M. Reed (1986) also states that further subdivision (e.g. moderately halotolerant and extremely halotolerant categories) is difficult because of the complexities of microbial responses to salt and osmotic pressures and other environmental parameters (e.g. temperature, pH, etc.) upon salt tolerances and salt requirements, as measured in the laboratory.

However they are classified, all microorganisms growing

in saline environments are faced with the problems of hyperosmotic conditions and the toxic effects caused by NaCl. Adaptations of halophilic and halotolerant organisms to their environments can be divided into three broad categories defined by Reed (1986) as being:

(1) Insulation. Adaptations that reduce the impact of the external environment upon the cell interior by preventing or limiting the entry of Na⁺ and the exit of water are in this category.

(2) Protection. In this category are mechanisms that protect intracellular function and metabolic activity from the inhibitory effects of Na⁺.

(3) Modification. This category includes those aspects of cellular metabolism which show optimal function in above normal salt conditions. Such processes show modification when compared to similar processes in salt sensitive cells.

These categories are not mutually exclusive. Halophilic and halotolerant organisms commonly employ a combination of adaptations from the categories listed above. Details for each category are given below.

Osmoregulation and cell membrane adaptations are categorized as insulation and should be considered a first line of defense. Growth in high salt environments is not so much a problem of solutes as it is of water availability (Brown, 1976; Kushner, 1978). Organisms living in such

environments are able to obtain and hold onto water via osmoregulation. It is recognized that halotolerance in microorganisms is directly related to the cell's ability to concentrate, to varying degrees, K^+ , Na^+ , and low molecular weight compounds intracellularly (Vreeland, 1987). Christian and Waltho (1961, 1962) showed that when Halococcus morruhae and Halobacterium cutirubrum were grown in 4.25 M NaCl, their cytoplasm contained 4.25 M KCl. The halophilic cyanobacterium Aphanothece halophytica as well as its halotolerant cousins also accumulate potassium ions as their primary mechanism of osmoregulation (Yopp *et al.*, 1978; Brock, 1976; Batterton and Van Baalen, 1971). Yopp *et al.* (1978) also discovered that the glycerol concentration in these organisms increased with increasing medium salt concentration. Eukaryotic algae and yeast have been found to concentrate polyols such as glycerol, mannitol, and arabinitol as well as potassium ions as osmoregulators (Fig. 1)

The marine algae Dunaliella tertiolecta and the halophilic algae D. viridis increase their glycerol concentration proportional to the extracellular salt increase (Borowitzka and Brown, 1974). Asteromonas gracilis is a halotolerant algae with a growth range that extends to 4.5 M NaCl for which it increases its glycerol content accordingly (Ben-Amotz and Grunwald, 1981). Reed (1980) has shown that the internal mannitol concentration of a halotolerant marine alga varies as a function of external salinity. The halotolerant and osmotolerant yeasts Debaromyces hansenii and

Fig. 1. Representative examples of organic osmoregulators:

(a) mannitol; (b) glycerol; (c) glutamate;

(d) glycine betaine; (e) proline.

PREVIEW